

## U.G. 4th Semester Examination 2022

## PHYSICS (Honours)

Paper Code : DC - 9

(Elements of Modern Physics)

Full Marks : 25

Time : Two Hours

*The figures in the margin indicate full marks.**Candidates are required to give their answers in their own words as far as practicable.*

1. Answer any *five* questions : 2×5=10
- (a) How the half life ( $\tau$ ) is related to the decay constant ( $\lambda$ ) of a radioactive sample ?  
Which value does ' $\lambda$ ' assume at the end product of a radioactive series ? 1+1
- (b) A hydrogen atom is  $5.3 \times 10^{-11}$  m in radius. Use the uncertainty principle to estimate the minimum energy an electron can have in this atom. 2
- (c) Which of the following is a possible solution of Schrödinger wave equation and why ?  
(i)  $A \cot x$  (ii)  $Ae^{-x^2}$ . 2
- (d) On fission,  $U^{235}$  yields two fragments of  $A = 95$  and  $A = 140$  roughly. Assume that the two fragments are ejected with equal and opposite momentum. Prove that their energies will be approximately in the ratio 3 : 2. 2
- (e) What do you mean by 'population inversion' ? How is it achieved in a ruby laser ? 1+1
- (f) The normalised wavefunction of a particle moving in a region  $0 \leq x \leq L$  is given by  
$$\psi(x) = \sqrt{\frac{2}{L}} \sin \frac{n\pi x}{L}$$
, where ' $n$ ' is an integer. Prove that the expectation value of the momentum of the particle is zero. 2
- (g) The nuclear radius of  $Be^8$  is 2.4 F. Find that of  $Al^{27}$ . 2
2. Answer any *three* questions : 5×3=15
- (a) Write down Planck's radiation formula, explaining the symbols used.  
Obtain Stefan's law of radiation from Planck's formula. 2+3

[P.T.O.]

- (b) (i) From the kinetic energy of photoelectrons versus frequency of the incident radiation, comment on the slope of the curve and indicate the threshold frequency. 2
- (ii) If  $u(x) = e^{-x^2/2}$  is an eigenfunction of the operator  $\left(\frac{d^2}{dx^2} - x^2\right)$ , find the corresponding eigenvalue. 1
- (iii) Show that uncertainty relation does not allow presence of electrons in the Nucleus. 2
- (c) (i) What are magic numbers ? Why are they so called ? 1+1
- (ii) Show from the semi-empirical mass formula, that  $A \approx 2z$  for light nuclei.
- Take  $\frac{a_c}{a_a} = 0.030$ . 3
- (d) (i) State Moscley's law of characteristic X-ray spectra. 1
- (ii) A particle of mass ' $m$ ' is confined to a one-dimensional box of length  $L$  extending from  $x = 0$  to  $x = L$ . Show that the probability of finding the particle in the region  $0 \leq x \leq L/4$  for an arbitrary value of quantum number ' $n$ ' is  $P = \frac{1}{4} - \frac{\sin \frac{n\pi}{2}}{2n\pi}$ . 4
- (e) (i) An electron of mass  $9.1 \times 10^{-31}$  kg is moving under a potential difference of 150 volt. Prove that the average wavelength of the corresponding de Broglie wave is nearly  $1.0 \text{ \AA}$ .
- (ii) Explain nuclear fission on the basis of liquid drop model. 2+3
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